The Refined Structure of the Complex between Adenylate Kinase from Beef Heart Mitochondrial Matrix and Its Substrate AMP at 1.85 Å Resolution

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(Received 8 August 1990; accepted 11 October 1990)

The crystal structure of the complex between adenylate kinase from bovine mitochondrial matrix and its substrate AMP has been refined at 1.85 Å resolution (1 Å = 0.1 nm). Based on 42,519 independent reflections of better than 10 Å resolution, a final *R*-factor of 18.9% was obtained with a model obeying standard geometry within 0.016 Å in bond lengths and 3.2° in bond angles. There are two enzyme : substrate complexes in the asymmetric unit, each consisting of 226 amino acid residues, one AMP and one sulfate ion. A superposition of the two full-length polypeptides revealed deviations that can be described as small relative movements of three domains. Best superpositions of individual domains yielded a residual overall root-mean-square deviation of 0.3 Å for the backbone atoms and 0.5 Å for the sidechains. The final model contains 381 solvent molecules in the asymmetric unit, $2 \times 72 = 144$ of which occupy corresponding positions in both complexes.

1. Introduction

Adenylate kinase from beef heart mitochondrial matrix (AK3[†], GTP: AMP-phosphotransferase, EC 2.7.4.10) is a monomeric enzyme that catalyzes the reaction $Mg^{2+}GTP + AMP \rightleftharpoons Mg^{2+}GDP + ADP$ (Tomasselli & Noda, 1979; Tomasselli *et al.*, 1979). The enzyme is essential for the biosynthesis of ADP from AMP. The peptide sequence has been reported by Tomasselli *et al.* (1986). The N terminus is heterogeneous, 33% and 18% of all molecules are shortened by one and two amino acid residues, respectively (Wieland *et al.*, 1984). A cDNA-derived sequence has also been published (Yamada *et al.*, 1989). The sequence homology to other members of the adenylate kinase family ranges between 24% and 37% identical residues (Schulz *et al.*, 1986).

Two medium-resolution crystal structures of adenylate kinases (AKyst, AKeco) with bound twosubstrate-mimicking inhibitor P^1 , P^5 -bis(adenosine-5'-)pentaphosphate (Ap₅A) and one high-resolution crystal structure of an adenylate kinase (AK1) without substrate have been reported (Egner *et al.*, 1987; Mueller & Schulz, 1988; Dreusicke *et al.*, 1988). Since AK3 was crystallized with only one of the substrates (AMP) bound, its crystal structure allows us to distinguish between the tri- and monophosphate position of bound Ap₅A. After reporting the initial X-ray analysis (Diederichs & Schulz, 1990), we now describe the final refined structure at a resolution of 1.85 Å (1 Å = 0.1 nm).

2. Methods

The crystal space group is $P2_12_12_1$ with 2 enzyme: substrate complexes per asymmetric unit. The unit cell dimensions are $49.83 \text{ Å} \times 67.99 \text{ Å} \times 155.41 \text{ Å}$ as derived from the analysis of rotation photographs. A native data set at 1.85 Å resolution (41,795 reflections) had been collected using the synchrotron source at the EMBL outstation in Hamburg (Diederichs & Schulz, 1990). After data reduction and merging with diffractometer data measured out to 3 Å resolution, a 95% complete data set at 1.85 Å resolution containing 42,869 reflections was produced (Diederichs, 1990) and used in all refinements. The refinement calculations were done with the molecular dynamics program X-PLOR (Bruenger et al., 1987), 1 cycle of which took up to 4 days of CPU time on our $\mu V\Lambda X$ -II. The first 6 rounds of the refinement resulted in a generally correct model with an R-factor of 28.4%, that has been reported previously (Diederichs & Schulz, 1990). The results of the subsequent refinement are given in

Table 1. In round 7 an additional alanine, Ala8, was

[†] Abbreviations used: AK3, adenylate kinase from beef heart mitochondrial matrix (EC 2.7.4.10); AKeco, adenylate kinase from *Escherichia coli* (EC 2.7.4.3); AKyst, adenylate kinase from bakers' yeast; AK1, adenylate kinase from pig muscle cytosol; Ap₅A, P¹,P⁵-bis(adenosine-5'-)pentaphosphate; NCS, noncrystallographic symmetry; r.m.s., root-mean-square.

 Table 1

 Structural refinement by simulated annealing in the resolution range 10 to 1.85 Å

Round ^a	R-factors (%)					
	min-1°	md-300°	min-2 ^d	min-B		
7	28.5	28.8	26.9	25.8		
8	25.9	27.3	25.1	24.3		
9			$24 \cdot 1$	23.4		
10	23.6	25.3	23.2	22.6		
11			22.2	21.7		
12	21.7	23.3	$21 \cdot 1$	20.7		
13	21.9	23.5	21.0	20.7		
14			20.8	20.6		
15	21.2	22.5	20.2	20.0		
16	20.3	21.9	20-0	20.0		
17			20.0	18.8		
18	19.2	19.9	19.0	18·9 ^f		

^aThe model was rebuilt after each round. At the end of each round the model geometry was always near ideality with r.m.s. deviations of 0.016 Å in bond lengths and 32° in bond angles.

^bFirst energy minimization (60 cycles), no NCS-restraint.

^cMolecular dynamics at 300 K for 0.25 ps, always NCS-restrained using a weight of 6 kcal/(mol· $Å^2$).

^d Second energy minimization (60 cycles), no NCS-restraint. ^e Temperature factor refinement (10 cycles), NCS-restrained for

rounds 7 through 16. ^f In round 18 the *R*-factor increased slightly due to deletion of 51 solvent molecules with low occupancies.

inserted before Ala9. It improved the fit of the preceding residues (Leu-Leu-Arg) to the electron density considerably (see Fig. 2). This correction was confirmed by repeating the N-terminal sequence determination of the protein (E. Schiltz, personal communication). Moreover, the additional alanine was also found in the cDNA sequence published recently (Yamada *et al.*, 1989). To stay consistent with previous publications, we decided to start sequence numbering with 0 instead of 1. This accounts somewhat for the inhomogeneous N terminus and it affects only the 8 N-terminal residues.

Most of the side-chains were positioned correctly after round 7. Minor errors in side-chain orientations, especially for Thr, Val, Leu, Asn, Gln, and His were corrected at later stages. For Asn, Gln and His and Thr we tried to establish the orientation that maximizes hydrogen bonding and that minimizes temperature factor discrepancies between the respective pairs of C, N and O atoms. The orientation of Leu and Val side-chains became unambiguous for temperature factors below about 30 Å².

Solvent molecules were usually identified in difference Fourier maps. They were added to the model if they were located at hydrogen bonding distance to the protein. If in later rounds the electron density of a solvent molecule dropped below a given threshold of about 80% of the standard deviation of the $(2F_{obs} - F_{calc})$ electron density map, it was removed from the model. When the *R*-factor crossed the 20% level at round 17 (Table 1), the temperature factors of NCS-related atoms were no longer restrained, causing another drop of 1.2%. The largest temperature factor differences of NCS-related atoms were observed in domain AMPbd (see Fig. 7).

As can be seen in Table 1, the refinement with X-PLOR proceeded smoothly. We did not apply molecular dynamics runs at 2000 K as proposed by Bruenger (1988) because the initial model was already generally correct. Instead, we kept the dynamics temperature at 300 K. Even at this temperature, the dynamics increased the radius of convergence of the refinement appreciably, because in all rounds the energy minimization following dynamics (min-2) yielded lower *R*-factors than the one preceding dynamics (min-1). The final *R*-factor was 18.9% for 42,519 reflections in the resolution range 10 to 1.85 Å. Here, one should keep in mind that the refinement included *all* measured reflections as observables. The *R*-factor for the subset of 29,074 reflections with intensities larger than 3 standard deviations is as low as 16.1%.

An error estimate for the atomic co-ordinates can be obtained from a Luzzati plot (Luzzati, 1952), which assumes that the disagreement between observed and calculated structure factor amplitudes is merely caused by atomic displacements. From Fig. 1, an upper error limit of 0.25 Å can be derived for the co-ordinate error. In this Figure we also included the *R*-factor for the subset of higher intensities, which lowers the error estimate.

Another error estimate is available from equivalent solvent positions in the 2 complexes. Two solvent molecules were considered as equivalent or NCS-related, if the distance between the 1st solvent molecule and the NCS-transformed position of the 2nd solvent molecule is less than 1 Å. Using this criterion and the NCS-relations separately for the domains MAIN, INSERT and AMPbd (Diederichs & Schulz, 1990), we observed 72 pairs of equivalent solvent molecules. The resulting distances were generally below 0.5 Å and the average distance was 0.3 Å although these pairs were not NCS-restrained during the refinement. This value agrees with the r.m.s. main-chain atom distance between complexes no. 1 and no. 2 after

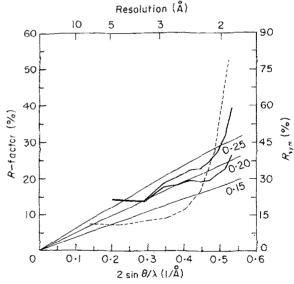


Figure 1. Luzzati plot for the final refined crystal structure of the AK3:AMP complex. The lines of constant co-ordinate error are labeled with the respective value. The upper continuous line (thick) represents R-factors of the final refinement for all 42,519 reflections in the range 10 to 1.85 Å resolution yielding an overall R-factor of 18.9%. The lower continuous line (thick) refers to R-factors calculated without the weak reflections (i.e. intensities smaller than 3 standard deviations) yielding an overall R-factor of 16.1%. The broken line gives the R_{sym} values derived from film data processing (scale at righthand side). In contrast to the R-factors, the R_{sym} values are calculated on intensities and can therefore not be compared directly. The R_{sym} values show, however, that the data quality decreases appreciably beyond 2 A resolution.

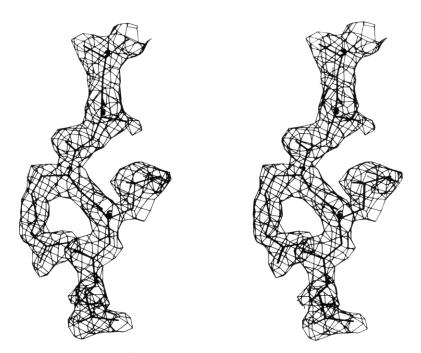


Figure 2. Electron density contour of a $(2F_{obs} - F_{cale})$ map together with the final model. Shown is segment Leu5-Leu6-Arg7-Ala8-Ala9 containing Ala8, which was initially derived from the electron density and is now confirmed by peptide and cDNA sequence analyses. The cut level is at 18% of the density maximum corresponding to 1.3 standard deviations of the map. The CA-atoms are labeled. For clarity, all density at distances of more than 2 Å from any atom of the model has been removed.

separate domain superposition. Considering these residual distances of main-chain atoms and solvent molecules (which are certainly augmented by local deviations from the NCS-relations) together with the Luzzati upper limit we estimate the actual co-ordinate error as about 0.2 Å.

3. Results and Discussion

(a) Main and side-chain conformations

As exemplified in Figure 2, the final electron density is of good quality. However, the five N-terminal (0 to 4) and the six C-terminal residues (220 to 225) are less well defined. The chain conformations at the N termini of complexes no. 1 and no. 2 are at variance, which is presumably caused by contacts to neighboring molecules. Residues 220 to 225 could only be positioned for complex no. 1 (into weak density); whereas no appropriate density was found in the case of complex no. 2.

The final model agrees well with the standard peptide geometry with r.m.s. deviations from ideality of 0.016 Å in bond lengths and 3.2° in bond angles. As shown in Figure 3, the main-chain dihedral angles fall mostly within the energetically favored regions (Ramachandran & Sasisekharan, 1968). Exceptions from this rule are found for three residues of complex no. 1: Ser2 at (ϕ, ψ) = $(-56^{\circ}, -79^{\circ})$, Gln219 at $(46^{\circ}, 26^{\circ})$ and Val223 at $(170^{\circ}, 95^{\circ})$, all of which are in regions of low density. In complex no. 2, Ser2 and Gln219 adopt conformations in allowed regions and Val223 is not included in the model. A general comparison of the (ϕ, ψ) angles of complexes no. 1 and no. 2 shows that for the structurally related residues 6 through 217 the r.m.s. deviation is 6° in ϕ as well as in ψ . This demonstrates the accuracy of these angles.

The geometry of the side-chains is given in Table 2. The expected trimodal distribution of the first side-chain dihedral angle χ_1 with a strong preference for staggered conformations g^- , t and g^+ is observed. There is almost no eclipsed conformation e. In a general way, the distribution agrees with those of other proteins as for example reported by Karplus & Schulz (1987) and McGregor *et al.* (1987). The statistics for complex no. 2 are similar. The r.m.s. χ_1 difference between the two complexes is 20°. This is three times higher than the r.m.s. difference of the main-chain angles and emphasizes the fact that the main-chain is better defined than the side-chains. For instance, the main-chain is less affected by crystal packing contacts.

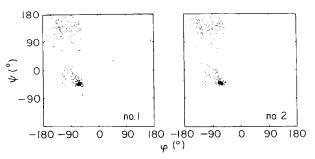


Figure 3. Scatter plot of all main-chain dihedral angles ϕ and ψ (Ramachandran & Sasisekharan, 1968) of the polypeptides of complexes no. 1 (left) and no. 2 (right). All glycine residues are omitted.

Residue – type		Nu	mber of resi	residues			Average values		
	N ^b	g^-	t	g^+	e	<i>g</i> ⁻	t	g^+	
Ser	10	3	1	3		57 (6)	162	289 (8)	
Thr	19	8	1	8		63 (5)	184	302(5)	
Val	15		12	2			173 (8)	301(5)	
Ile	12		1	11	1		186	287 (12)	
Asp	12	1	4	6		60	185 (5)	294(5)	
Asn	8	1	3	3		62	174 (4)	286 (13)	
Glu	15		3	7			201(15)	298(7)	
Gln	10	1	1	5	1	60	185	291 (7)	
Met	4		1	2			191	287(1)	
Lys	14		3	8	1		177(4)	300 (17)	
Arg	16		4	10			190 (13)	287 (10)	
Leu	26	2	9	15	_	94 (1)	185 (5)	294 (7)	
His	5		3	2			178 (3)	297(1)	
Phe	7		3	3			179(4)	291(4)	
Гуr	7		3	4			181 (8)	293(4)	
Trp	3	1	1	1		30	173	290	
Total	183	17	53	90	3	63 (14)	181 (11)	293(11)	
Total-α°	77	3	30	39	2	58 (6)	181 (12)	292 (8)	
$\operatorname{Total}_{\beta^d}$	25	1	12	12	1	30	180 (11)	293 (10)	
Total-non- $\alpha\beta^e$	81	13	11	39	manutur.	67 (14)	181 (11)	293 (12)	

Table 2Statistics of side-chain torsion angles χ_1 of complex no. 1^a

^a χ_1 is the torsion angle at the CA–CB bond. For exact definition of χ_1 as well as g^- , t, g^+ and e see Karplus & Schulz (1987). Standard deviations are given in parentheses.

^b Number of residues present in the sequence. For 20 residues at least 1 of the temperature factors of the CA, CB or CG atoms exceeded 60 $Å^2$ so that no conformation was assigned.

 $^{\circ}$ Residues in α -helices.

^d Residues in β -sheets.

^e Residues neither in α -helices nor β -sheets.

(b) Secondary structures and hydrogen bonds

The secondary structure assignments were made using the simple cut-off criterion (distance between H- and acceptor atom smaller than 2.5 Å, regardless of the angle of approach) suggested by Baker & Hubbard (1984). The assignments are given in

0	1	2	3	4	5	6	7	8
0	0	0	0	0	0	0	0	0
h-GASA	ARLLRAAIM	CAPGSGKGTVSSRI	FKHFELKH	LSSGDLLRDN	LRGTEIGVI	LAKTFIDQGK	LIPODVMTRLV	LHELKNL
s	S EEEEEE	тт никнични	нннн ее:	ееннынныны	инсниния	нннннядс	нниннини	нниннас
	EEEEEE	тт нинниннын	нини ее:	ееннининни	нннкники	ннннндс	нниннин	нннннсс
	par	helix-1	pa	r helix-2	heli	c− 3	helix-4	
	9	10	11	12	13	14	15	16
	0	0	0	0	0	0	U	0
TQS	INWLLDGFP	RTLPQAEALDRAYQ	TOTATIVEN	VPLEVIKORL.	ARWINPGSC	SKVINIEFNP	PRIMGIDDLIG	EPLVQRE
TŤ	EEEEESSS	HHHHHHHHH	EEEEEE	нннкнинин	IEEEEETTT	EEEEETTS	SSTEEETTT	EEEE GG
TT	EEEEESSS	ннининнинт	EEEEEE	ннннннн	EEEEETTT	EEEEETTS	SSTEEETTT	EEEE GG
	par	helix-5	par	helix-6	anti	anti	anti	anti
					21	22		
	17	18	19	20	21	22		
DDT		AYEAQTEPVLEYY		v				
					-		V1F-011	
GG)	ннннннн	иннинниннинни	HHHEEEEI	еее тттння	ннннннтт	r s	s	
GGE	нннннн	книнникнинни	HHGEEEE	сее тттнн	нннннетт	r		
	helix-7	helix	-8 pa	ar he	lix-9			

Figure 4. Amino acid sequence and secondary structure assignment of AK3. Since the model is well refined, we used the rather restrictive cut-off criterion (distance between H and acceptor atom smaller than 2.5 Å) and followed Karplus & Schulz (1987). From top to bottom the lines contain: sequence, secondary structure of complex no. 1, secondary structure of complex no. 2, labels at helices and at the *parallel* and *antiparallel* β -sheets of domains MAIN and INSERT, respectively. The assignment is: H, α -helix; E, β -pleated sheet; G, 3₁₀-helix; T, loop with an (i+n, i) hydrogen bond (n = 3, 4, 5); S, loop without hydrogen bond. Figure 4. They differ only slightly for the two polypeptides. The basic structural motif of all adenylate kinases consists of a five-stranded, parallel β -sheet (Fig. 5) surrounded by nine α -helices, where helices 7 and 8 are separated by a proline, here Pro180. This proline is conserved in all adenylate kinases. It disturbs the hydrogen bonding pattern so that helix 7 changes its direction by about 30° before continuing as helix 8.

While this part of the structure resembles the chain fold of the small variants of adenylate kinases (Schulz *et al.*, 1986), a stretch of 39 residues forms the separate domain INSERT consisting of two β -meanders. These β -meanders are loosely connected to form a six-stranded curved antiparallel β -pleated sheet specified in Figure 5 and illustrated in Figure 6. This domain is further stabilized by side-chain hydrogen bond interactions as listed in Table 3.

The statistics of main-chain dihedral angles and hydrogen bond parameters in α -helices and β -sheets are given in Table 4. They are in general agreement with those of other proteins (Karplus & Schulz, 1987).

(c) Non-crystallographic symmetry

The two complexes in the asymmetric unit are related by NCS (residues 0 to 4 and 220 to 225 are here neglected), which was exploited for density averaging during the initial phases of structure

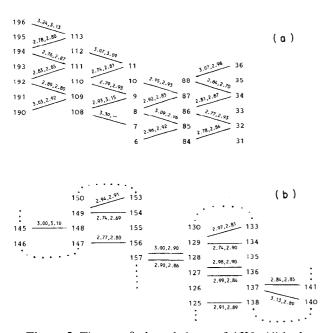


Figure 5. The two β -pleated sheets of AK3. All hydrogen bond lengths of complexes no. 1 and no. 2 are given as donor... acceptor distances. (a) Central parallel sheet of domain MAIN observed in all adenylate kinases. (b) Antiparallel sheet of the domain INSERT, which is only present in the large variants among the adenylate kinases (Schulz *et al.*, 1986).

elucidation. Soon after beginning the refinement, significant deviations from an overall NCS relation were recognized (Diederichs & Schulz, 1990). They amounted to up to 3 Å for backbone atoms and gave rise to an r.m.s. deviation of $1\cdot 1$ Å (all atoms) between the two complexes. This number drops appreciably if the domains MAIN, INSERT and AMPbd are separately overlaid. In this case, the all-atom r.m.s. deviation is $0\cdot 4$ Å. If subdivided into main- and side- (given in parentheses) chain atoms

 Table 3

 Hydrogen bonds involving side-chains of the domain

 INSERT

		Donor–acceptor distance in complex		
Atom-1	Atom-2	no. 1 (Å)	no. 2 (Å)	
Arg134–NE ^a	Asp149-OD1	2.88	2.89	
Arg134-NH1	Asp149–OD2	2.82	2.97	
Val135–O	Asn141–ND2	3.30	3.14	
Tyr136–OH	Asp149–OD1	2.75	2.76	
Tyr136–OH	Asp150-N	2.98	3.20	
Åsn137–N	Asn141-OD1	2.69	3.02	
Lys144-N	Asp150–OD2	2.97	2.89	
Asp149-OD2	Thr152-N	3.18	3.07	

^a This is a salt bridge at the surface.

the r.m.s. deviations are: 0.32 Å (0.52 Å) for MAIN (residues 5 to 32, 59 to 123, 163 to 215), 0.38 Å (0.52 Å) for AMPbd (residues 33 to 68) and 0.27 Å (0.50 Å) for INSERT (residues 124 to 162).

The deviations are best characterized by superimposing domain MAIN of complex no. 1 on its counterpart in complex no. 2 and comparing the resulting positions of the other domains. This superposition shows that in complex no. 1 domain INSERT is rotated by 9° toward domain AMPbd and that domain AMPbd is rotated by 7° toward INSERT. Both rotations close the large substrate binding cleft (Diederichs & Schulz, 1990) to some extent, rendering complex no. 1 more "closed" than no. 2. These rotations are almost certainly due to the different contact environments in the crystal. They also demonstrate that these domains are not rigidly glued together. Apart from the domain rotations there are a number of side-chains with differing conformations, most of which are at the surfaces.

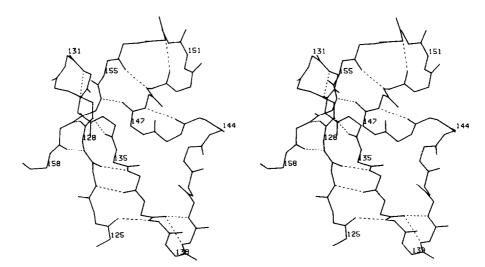


Figure 6. Main-chain conformation of domain INSERT of AK3:AMP complex no. 1 (complex no. 2 is closely similar). The chain contains 2 consecutive β -meanders that form a curved 6-stranded antiparallel sheet. If residues 141 and 145 are considered as being in the same β -strand, the curved 6-stranded sheet can be described as a 5-stranded barrel.

 Table 4

 Average main-chain dihedral angles and hydrogen bond parameters in α-helices and β-sheets^a

			$\beta ext{-Sheet}$		
	α-Helix	Parallel	Antiparallel	Total	
$\overline{\langle \phi \rangle^{\mathfrak{b}}}$	-64(8)	-113	-110	-112 (18)	
$\langle \psi \rangle^{\circ}$	-40(8)	130	140	133 (15)	
$\langle N \dots O \rangle^d$	3.00(0.15)	2.92	2.90	2.91(0.13)	
$\langle \mathbf{H} \dots \mathbf{O} \rangle^{\mathfrak{e}}$	2.06 (0.16)	1.98	1.97	1.97 (0.15)	
$\langle N - \hat{H} \dots O \rangle^{f}$	158 (10)	159	156	158 (11)	
$\langle \mathbf{H} \dots \hat{\mathbf{O}} = \mathbf{C} \rangle^{\mathbf{g}}$	151 (8)	156	148	153 (13)	

⁴ The secondary structures are as defined in Fig. 4. Standard deviations are given in parentheses.

^b Average angle ϕ (°).

^c Average angle ψ (°).

^d Average distance N . . O (Å).

^e Average distance H_.. O (Å).

^f Average angle $N - \hat{H} \dots \hat{O}$ (°).

⁸ Average angle H . $\hat{O} = C$ (°).

(d) Temperature factors

The mean temperature factor for all atoms of the refined model is $33\cdot0$ Å². Subset averages are: $26\cdot5$ Å² for the main-chain atoms, $38\cdot3$ Å² for sidechain atoms and $49\cdot2$ Å² for solvent atoms. The variation of the temperature factors along the mainchain is plotted in Figure 7. A striking feature in this Figure is the high temperature factor of domain AMPbd of complex no. 1 in comparison to complex no. 2. This difference can be explained by the crystal lattice contacts (see below). Towards the N and C termini, the temperature factor rises sharply in both complexes. It seems therefore likely that neither N nor C termini have defined structures in solution.

The occupancies of all solvent molecules have been fixed at 100%. The mean temperature factor of the NCS-related solvent molecules modeled as water is merely 40.0 Å² while that of the other water molecules is 55.0 Å², which indicates a tighter

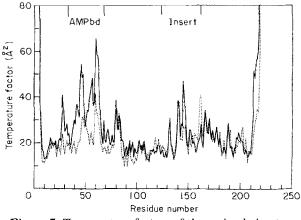


Figure 7. Temperature factors of the main-chain atoms (averaged per residue) of AK3:AMP complex no. 1 (continuous line) and complex no. 2 (dotted line). Domains AMPbd and INSERT are indicated.

		Table 5		
Contacts	between	AK3:AMP	complexes	in the
		$crystal^{a}$		

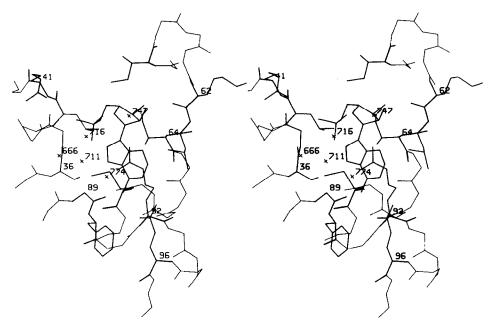
		Ну	Hydrogen bonds ^d			
Packing interaction ⁶	Buried area (Å ²) ^c	Ref. set ⁵	2nd set ^b	DA distance (Å) ^d		
I	550	201-0 205-NE2	62–NZ 62–NZ	3·44 2·85		
II	383	59-OD1 158-N 158-O 160-OE1 160-OE2 163-NE 163-NH1	201-NZ 211-OE1 211-NE2 203-NE1 203-NE1 207-OH 219-NE2	2·95 3·00 2·84 2·93 3·37 3·04 3·18		
III	399	49–N 49–OE2 75–NE2	30-OE2 217-NH2 219-O	3·06 2·87 2·79		
IV	514	7-NE 7-NH1 101-OD1 101-O 103-O 105-N	132-0 132-0 126-NH1 126-NH1 135-N 133-0	3·02 3·19 3·07 2·77 2·96 2·84		
V	504	41-NE 126-NH2 126-NH2 132-O 133-O 133-O 134-NH2 135-N	187-O 105-OE1 105-NE2 7-NE 105-N 104-OH 103-O	3·46 3·49 2·74 3·39 3·03 3·44 3·04		
VI	216	116–N 117–N 175–OE2	151-0 151-0 144-NZ	$3.42 \\ 2.90 \\ 2.77$		
VII	181	None				
VIII	57	None				
IX	100	None				
Х	172	0-O 0-N 1-N 2-N	192–OE2 189–O 192–OE2 192–OE1	3·06 2·86 2·65 3·08		

 $^{\rm a}$ The given contacts contain at least one interatomic distance below 4.5 Å.

^b The packing contacts I through X are formed by a residue set of complex no. 1 (res-1) or complex no. 2 (res-2) taken as a reference (operator = [X, Y, Z]) and a second residue set of another complex (res-1 or res-2) related by the operator given as OP[...]. I. res-1 (200-205, 207, 208, 211, 216, 221, 225) with OP[X-1, Y, Z] on res-2 (44, 45, 47, 49, 52, 53, 56, 57, 60, 62, 65). II: res-1 (59, 155, 157-160, 163-165) with $OP[\frac{1}{2} - X, -Y + 1]$, $\frac{1}{2}+Z$ on res-2 (28, 201, 203, 207, 208, 211, 216, 219). III: res-1 (45-49, 52, 75) with $OP[\frac{3}{2}-X, -Y+1, \frac{1}{2}+Z]$ on res-2 (0, 2, 3, 5, 30-32, 82-84, 215, 217-219). IV: res-1 (2, 3, 5, 7, 81, 101-105, 183, 186, 187) with OP[$X + 1, \frac{1}{2} + Y, \frac{1}{2} - Z$] on res-2 (38, 41, 45, 126, 128, 131–135, 141, 159). V: res-1 (38, 41, 126, 131–135, 141) with $OP[-X+1, \frac{3}{2}+Y, \frac{1}{2}-Z]$ on res-2 (5, 7, 81, 101-105, 186, 187). VI: res-1 (115–117, 175) with OP[$-X, \frac{1}{2} + Y, \frac{1}{2} - Z$] on res-2 150-152). (144, VII: res-1 150 - 153)(144,with $OP[-X, -\frac{1}{2} + Y, \frac{1}{2} - Z]$ on res-2 (78, 115–118, 197). VIII: res-1 (78, 81) with OP[X-1, Y, Z] on res-1 (117, 120). IX: res-2 (46. 78) with OP[X+1, Y, Z] on res-2 (117, 121, 145, 146). X: res-2 (0-2) with $OP[\frac{1}{2}+X, -\frac{1}{2}-Y, -Z]$ on res-2 (189-192, 209).

^c As determined using program DSSP of Kabsch & Sander (1983).

^aAll hydrogen bonds within the contacts with donor... acceptor distances less than 3.5 Å are listed.



(a)

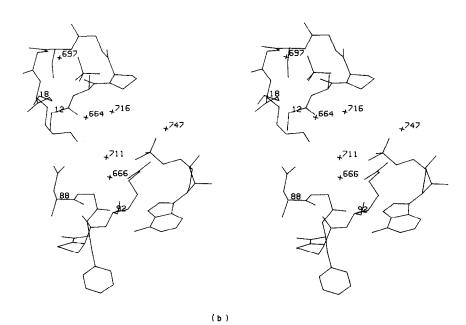


Figure 8. The environment of bound AMP in AK3:AMP complex no. 1 (complex no. 2 is very similar). For clarity, all side-chains that do not contact AMP are deleted. Several CA-atoms and water molecules are labeled. (a) The immediate environment of AMP. (b) The geometric relation between the phosphate of AMP and the sulfate molecule (upper left-hand side) bound at the giant anion hole formed by the glycine-rich loop in a position similar to AK1 (Dreusicke & Schulz, 1986).

binding of the NCS-related water molecules. They can be considered as an integral part of the protein. In both complexes, the temperature factor of the bound sulfate is around 80 Å². These sulfates occupy the "giant anion hole" formed by the glycine-rich loop as observed in AK1 (Dreusicke & Schulz, 1986), but they are much less rigorously bound than in AK1.

(e) Crystal packing

As demonstrated in Table 5, the strongest packing interactions in the crystal occur between the NCS-related complexes. The contacts between complexes related by crystallographic symmetry are much weaker. Ala103–Gln105 and Val135–Gly113 form a small distorted intermolecular antiparallel

AMP atom	Protein atom	Distance (Å)	H A (Å)	$\mathbf{D} - \mathbf{\hat{H}} \dots \mathbf{A} (^{\circ})$
NIB	Gln96-NE2	2.91 (2.99)	1.95 (2.03)	166 (163)
NIB	Met69–CG	3.66 (3.66)		
N3B	Ile64-N	3.25(3.31)	2.35(2.43)	153 (150)
N3B	Leu63–CD2	3.58 (3.67)		
N7B	Ser36–OG	2.92(2.92)	1.97 (1.97)	165 (163)
N7B	Arg92–NH2	3.65 (3.08)	3.34 (2.63)	110 (107)
N7B	Gly37–N	4.29(4.42)	4.09 (4.26)	95 (93)
N9B	Leu40-CD1	3.52 (3.58)	· · /	· · ·
N10B	Gln96–OE2	2.86 (3.02)	1.89 (2.07)	160(155)
N10B	Gly89–O	2.87(2.85)	1.89(1.87)	162 (160)
N10B	Phe90-CA	3.89 (3.73)		(/
OIRB	Leu40–CD1	3.66 (3.28)		
02RB	Lvs62–O	2.54 (3.06)	1.78(2.91)	134 (89)
O3RB	Wat782	2.93	2.55	102
O3RB	Wat958	(4.12)	(3.78)	(103)
C4RB	Ile58–CG1	4.12 (4.16)	~ /	
O5RB	Gly37–CA	4.23 (3.63)		
OP5R ^b	$\tilde{Wat747}$	2.50	1.85	120
OP5R	Wat901	(2.41)	(1.86)	(111)
OP5R	Arg92–NH1	3.85 (4.14)	3.00 (3.25)	143 (150)
OP5R	Wat716	4.34	3.74	121
$OP5S^{b}$	Wat711	2.50	1.87	118
OP5S	Wat903	(3.09)	(2.16)	(154)
OP5S	Wat666°	3.33 (2.55)	3.05 (1.83)	97 (126)
OP5S	Arg92–NH2	3.53 (4.37)	2.64(3.44)	142 (155)
OP5S	Arg92-NH1	3.50 (4.76)	2.60 (3.92)	149 (143)
OP5S	Wat774	4.14	4.07	87
045 ^b	Arg41–NH1	3.53(2.51)	3.27 (1.97)	83 (111)
D45	Wat666°	3.63 (4.28)	2.68 (3.30)	159 (167)
045	Wat918	(4.41)	(3.68)	(132)
045	Wat928	(4-22)	(3.39)	(142)
045	Gly37–CA	3.91 (4.03)	· · ·	· · /

 Table 6

 Contacts between atoms of the substrate AMP and the enzyme^a

 a All contacts with distances smaller than 4.5 Å in any one of the complexes are listed. The values for complex no. 2 are given in parentheses following those of complex no. 1.

^b The oxygen atoms OP5R, OP5S, O45 are named as in Ap₅A. The numbers denote the phosphate number (1 through 5) in Ap₅A. ^c The values given for complex no. 2 are those of the NCS-related solvent molecule Wat866.

 β -sheet. This contact occurs twice (packing interactions IV and V of Table 5) such that it does not give rise to dimers, but connects subsequent molecules forming a linear string along the *b* axis of the crystal. These favorable contacts along the *b* axis are likely to enhance crystal growth along this very direction. This corresponds to the observation that crystals are usually about ten times larger along the *b* axis than along the *a* and *c* axes.

The lattice contacts of domain AMPbd differ appreciably in the two complexes. In complex no. 1, AMPbd contacts only a few residues of an NCS-related molecule, whereas AMPbd of complex no. 2 is stabilized by a contact with the C-terminal helix of complex no. 1. As a consequence, the average temperature factor for AMPbd is lower in complex no. 2 (28 Å²) than in complex no. 1 (40 Å²). The same applies for the mean temperature factor of the bound AMP (29 Å² in complex no. 1, 40 Å² in complex no. 2).

(f) The AMP binding site

The AMP binding site is located in the pocket between the helices 2, 3 and 4. The contacts of AMP are given in Table 6. The binding geometry is illustrated in Figure 8. In crystals of AKyst:Ap₅A and AKeco:Ap₅A, one of the adenines binds at the equivalent location. The known high specificity of the adenylate kinases towards AMP, which contrasts their lower specificity for ATP, together with the highly conserved sequence in this region allowed Egner *et al.* (1987) to assign this site as the AMP site. This hypothesis is confirmed by the crystal structure presented here.

Hydrogen bond interactions between enzyme and substrate AMP involve mostly the nucleotide base of AMP; all nitrogen atoms take part. Adenine is hydrogen-bonded to the side-chains of Ser36 and Gln96 as well as to the amide of Ile64 and the carbonyl of Gly89. Gly89 is located in a loop which interacts with the Gly-rich loop via salt bridge Asp88-OD2:Lys18-NZ. The loop conformation is stabilized by cis-Pro91, which is conserved in all known adenylate kinases. While Gln96 is also a conserved residue, Ser36 is exchanged for Thr in some species (Schulz et al., 1986). But Thr has a hydroxyl in the y-position like Ser, so that this exchange has probably only minor effects. One may therefore assume that these four residues warrant the high specificity of the adenylate kinases for the base adenine.

The ribose and the phosphate of AMP are much less rigidly bound than the base. The average temperature factors for (adenine, ribose, phosphate) are (25 Å², 45 Å², 63 Å²) in complex no. 1 and (17 Å², 31 Å², 48 Å²) in complex no. 2. Only the 2'-hydroxyl of the ribose forms a strong hydrogen bond directly to the polypeptide; all other ribose atoms are either hydrogen-bonded to water molecules or they contact non-polar atoms (Table 6). The phosphate contacts water as well as the guanidinium groups of Arg41 and Arg92.

Taken together, the base adenine seems to sit snugly at its final position for catalysis while further movements have to be expected for ribose and phosphate. As shown in Figure 8(b), the sulfate bound in the glycine-rich loop is rather far away from the phosphate of AMP. Since this sulfate corresponds to the β -phosphate of ATP as derived from Ap₅A bound to the adenylate kinases AKyst as well as AKeco (U. Abele & C. W. Müller, personal communications), it is clear that some movement of AMP relative to ATP has to occur to allow transfer of the γ -phosphoryl group of ATP. Such a movement has been derived from a comparison between several different adenylate kinase structures (Schulz *et al.*, 1990).

We thank Dr A Tomasselli, Dr L. H. Noda and S. Burger for providing us with the enzyme and the crystals for this study.

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Edited by R. Huber